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Chapter 7 - Chemical Formulas and Chemical Compounds 7-1  
Chemical Names and Formulas I. Significance of a Chemical Formula A. Molecular formulas 1. Number of atoms of each element in one molecule of a compound  $C_2H_6 =$  ethane (2 carbon atoms, 6 hydrogen atoms) B. Ionic Compounds 1.

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Modern Chemistry 69 Chapter Test Chapter: Chemical Equations and Reactions PART I In the space provided, write the letter of the term or phrase that best completes each statement or best answers each question. \_\_\_\_ 1. The production of a slightly soluble solid compound in a double-displacement reaction results in the formation of a a. gas. b ...

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CHAPTER 7 REVIEW Chemical Formulas and Chemical

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Compounds MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Write formulas for the following compounds: CuCO<sub>3</sub> a. copper(II) carbonate Na<sub>2</sub>SO<sub>3</sub> b. sodium sulfite (NH<sub>4</sub>)<sub>3</sub>PO<sub>4</sub> c. ammonium phosphate SnS<sub>2</sub> d. tin(IV) sulfide HNO<sub>2</sub> e. nitrous acid 2.

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Match each item with the correct statement below. a. monatomic ion f. cation b. acid g. binary compound c. base h. anion d. law of definite proportions i. polyatomic ion e. law of multiple proportions \_\_\_\_ 1.

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Chemistry Chapter 7 Test Multiple Choice Identify the choice that best completes the statement or answers the question. \_\_\_\_ 1. A chemical formula includes the symbols of the elements in the compound and subscripts that indicate a. atomic mass of each element. b. number of atoms or ions of each element that are combined in the compound. c ...

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